AUTOCHEM III

A Catalyst Characterization

Laboratory in a Single Analytical Instrument

micromeritics.com/AutoChem





The AutoChem from Micromeritics is the most widely used and highly cited system for catalyst reactivity characterization because it is also the most automated, highly accurate system for chemisorption and temperature-programmed reactions.

The all new AutoChem III meets and exceeds that performance with a design that will save you hours a day, make the most sensitive, reproducible measurement, and enhance operator safety.

Pulse Chemisorption	Metal Dispersion	Metal-Supported Catalysts
TPR	Metal Surface Area	Acid or Base Catalyzed Reactions
TPO	Active Surface Area	Oxide or Zeolite Catalysts
TPD	Crystallite Size	Advanced Battery Anode Materials
TPSR	Heat of Desorption	Fuel Cell Catalysts
Dynamic B.E.T.	Activation Energy	
Breakthrough Curve	B.E.T. Surface Area	

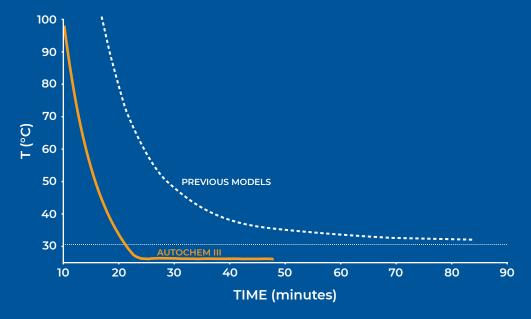


QUICK & EASY

The new AutoChem III is designed to make key operations quick and easy, saving you hours a day so you spend less time making measurements and more time making progress.

RAPID TURNAROUND TIME WITH AUTOCOOL

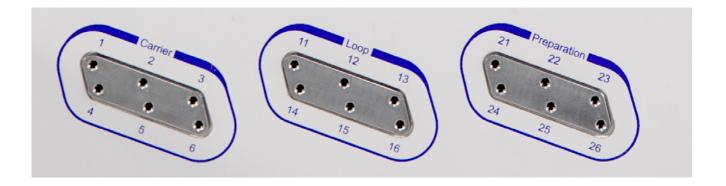
The NEW AutoCool is an integrated gas-fed system that rapidly cools sample tubes before and during experiments. AutoCool is typically **30 minutes faster** than alternative systems and requires no liquids or external support.





NEVER PREPARE ANOTHER VAPOR CAPTURE SLUSH BATH

The NEW AutoTrap effectively traps vapor and requires **no manual slush bath preparation**. Traditional methods for vapor capture require the time-intensive process of slush-bath preparation through the manual mixing of liquid nitrogen with isopropanol. The zeolite bed of the AutoTrap effectively captures vapors, can be used for several experiments without interruption, and can be regenerated *in situ*.



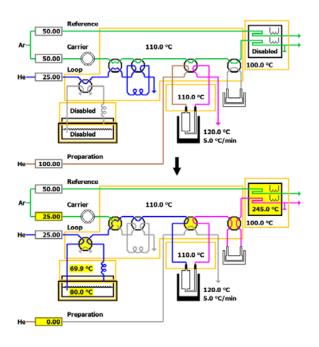
READY TO RUN WITH 18 AVAILABLE GAS STREAMS

Don't waste time reconnecting and switching gas lines: have what you need ready when you need it. The AutoChem III has 18 available gas streams so you're always ready to run your next reaction.

Having the right blended gas ready also means you won't introduce errors from poorly designed external gas connections, and you won't compromise data accuracy by blending gases, which unnecessarily introduces error from mass flow controllers.

PROGRAM WHAT YOU ENVISION, VISUALIZE WHAT YOU HAVE PROGRAMMED

The new MicroActive method editor features an intuitive process illustration that shows the programmed state of the instrument at every step of the method so you can see that your method matches your vision.

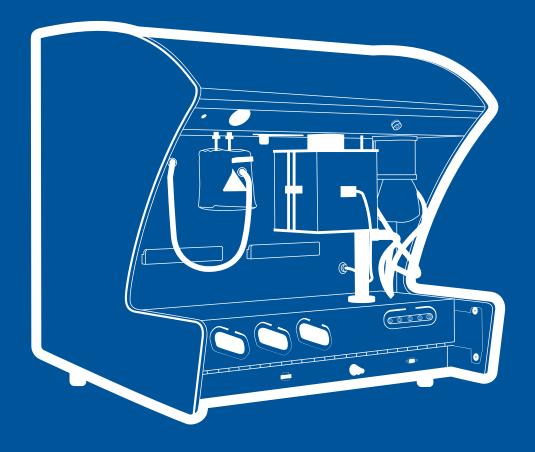


MAKING ACCURACY EASY: EXCLUSIVE AUTOMATED DETECTOR CALIBRATION

The AutoChem III makes accuracy simple with automated detector calibration. Traditional systems require calibration by multiple runs of reference materials or single-point offsets that ignore changes in temperature or pressure.

The AutoChem III makes the most accurate gas volume measurement through an automated multi-point calibration using patented gas blending and a dedicated reference stream flow controller. Injection loop temperature control and pressure measurement reduce error by an additional 5% compared to inferior designs. The process is **fast, automated, requires no operator intervention, and produces more accurate results than alternative designs.**

US Pat #10,487,954 B2





LOADING SAMPLES IS A SNAP

The patented new KwikConnect makes sample tube installation faster, easier, and more reliable than traditional designs with half as many separate pieces and no threaded fittings. Installation and removal is easier and quicker, reduces the risk of breaking sample tubes, and provides peace-of-mind that the snap lock closure has completely sealed the system.

6 QUICK & EASY US PAT #11,105,825 B2





Internal gas temperature control

mi micromeritics

in four separate zones prevents condensation during studies with vapor and improves overall signal stability

KwikConnect

retention system makes sample tube mounting quick, easy, and safe with no threaded connections and half as many separate pieces as traditional designs

Exclusive AutoTrap

provides superior moisture removal for TPR experiments with a system that is effortless to use and saves hours per day

18 total gas inlets

six each for preparation, carrier, and loop gases permit sequential experiments of different types and saves time between experiments

Lowest internal gas volume

provides the highest peak resolution and minimizes tailing when changing gas stream composition.

High sensitivity thermal conductivity detector (TCD)

is 2x as sensitive as alternatives so you can measure smaller samples, detect secondary reactions, and have greater confidence in your results

Temperature-controlled corrosion-resistant detector

is compatible with corrosive gases and inherently protected from gas leaks that can destroy alternative designs, providing high reliability and long operational life

Integrated AutoCool

cools the furnace and sample quicker than forced air alone without the use of cryogenic liquids, saving an average of 30 minutes per experiment

Dynamic clamshell furnace

provides temperature control up to 1200°C and controlled heating rates from 0.1°C/min to 100°C/min with the lowest available temperature overshoot



BETTER MEASUREMENTSFOR MORE CONFIDENT DECISIONS

The AutoChem III provides results that drive confident decisions. The highest available measurement accuracy and repeatability - made under conditions that match your reaction environment - give you certainty to act with confidence.

PRECISION TEMPERATURE CONTROL

Exacting thermal accuracy is essential to simulate reaction conditions while preventing deactivation of precious catalyst materials. The AutoChem III exceeds all available systems in every key performance characteristic

- · Widest range of temperatures: -100°C to 1200°C
- Widest range of heating rates: 0.1 °C/min to 100 °C/min Repeatable temperature profiles
 Accurate determination of activation energy, E_a
- Local sample temperature measurement
 Superior control accuracy without overshoots
- Four independently controlled gas stream temperature zones eliminate vapor condensation and enhance measurement stability

MOST ACCURATE GAS STREAM COMPOSITION

The AutoChem III features the lowest gas flow path volume to **eliminate carryover** and signal tailing when changing gas flow conditions. This guarantees precise gas stream composition, even when switching configurations from one experiment to another.

And with 18 available gas inlets, you will have the gas composition that you need ready, without introducing errors associated with blending gases *in situ*.

BETTER TEMPERATURE CONTROL AT EVERY STEP

FURNACE

to simulate reaction conditions

VAPOR

to control to maximize vapor composition detection sensitivity

GAS STREAM

to ensure robustness

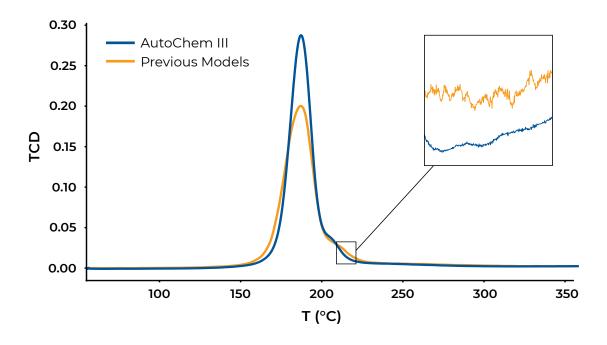
DETECTOR

SEE MORE OF YOUR REACTION WITH THE MOST SENSITIVE CHEMISORPTION SYSTEM IN THE WORLD

The new AutoChem III features a new thermal conductivity detector (TCD) that is 110% more sensitive than previous designs. This enables you to use lower sample masses, accurately detect secondary reactions, and achieve greater accuracy of catalyst traits like site coverage.

Detector sensitivity is enhanced by a reference stream with a dedicated mass flow controller (MFC) that provides a stable reference to the sample stream. Alternative designs use a common carrier stream for both the reference and signal paths, resulting in interference between the measurement and reference stream leading to signal instability.

The temperature-controlled TCD is a robust sensor with a long operating life and intrinsic protection from operational errors like gas flow leaks that cause premature failure of 4-element detectors used in inferior designs.





CONTINUOUS CONTROLLED VAPOR DOSING

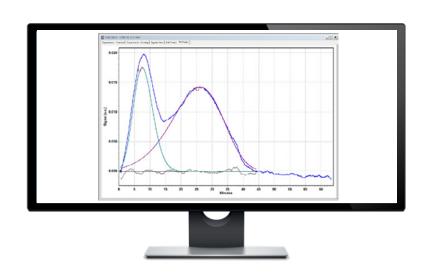
Achieve faster analysis and more complete characterization of surface selectivity and functionality with the available vapor generator featuring automated vapor calibration, injection repeatability better than 1%, and all new continuous dosing capabilities. This system creates uniform streams of saturated vapors such as water, alcohols, amines, or organics that are used to prepare samples for TPD or as the reaction gas stream.

The new continuous dosing capability enables faster and more uniform vapor dosing than legacy systems that are limited to discrete vapor stream pulses.

MOVE QUICKLY FROM DATA TO DECISION

Rapidly transition from experimental data to material characteristics with Micromeritics' own AutoChem data analysis software. Get all the answers you need with:

- Interactive peak analysis including: limit selection, baseline definition, integration, and deconvolution
- Built-in analysis models for: pulse chemisorption, % dispersion, metal surface area, crystallite size, First-order kinetics, heat of desorption, activation energy, BET, Langmuir, total pore volume, and more.
- Seamless integration of mass spec data
- Detailed, configurable graphical reports



IMPROVING OPERATOR SAFETY

The AutoChem III enhances operator safety at every stage of the measurement, reducing opportunities for exposure and potential for hazardous conditions.

NO CRYOGENIC LIQUIDS

The new AutoTrap removes moisture without cryogenic liquids such as liquid nitrogen. The AutoTrap also eliminates the need for slush bath preparation, which requires vigorous mixing of alcohols and other solvents in glass vacuum flasks.

COOL TO THE TOUCH

The new AutoCool brings sample tubes to room temperature so fast after every experiment that you can change samples and start the next experiment quickly, without handling hot glass sample tubes. And the KwikConnect sample tube retention system allows you to release the tube in one motion without fumbling with threaded connections and separate adapter pieces.

THIRD-PARTY TESTED AND VERIFIED

Micromeritics products are third-party-tested to conform to the highest level of regulatory compliance and operational safety. Install and run with the confidence that the system will meet or exceed requirements for electrical safety and compatibility with the need for separate qualifications or assessments.









INSTRUMENTCAPABILITIES

AutoChem III

Temperature	Ambient to 1200°C		
Temperature Ramp Rates	-100°C to 800°C: up to 100°C/min 800°C to 1000°C: up to 50°C/min 1000°C to 1200°C: up to 25°C/min		
Preparation gases	6 inlets: H ₂ , O ₂ , He, Ar, H ₂ /Ar, and more		
Carrier gases	6 inlets: He, Ar, H ₂ /Ar, and more		
Analysis (loop) gases	He, H ₂ , CO, O ₂ , N2O, NH ₃ /He, and more		
Pulse Chemisorption			
Temperature-programmed reactions: TPR, TPO, TPD, TPSR			
Strong Chemisorption: Reactive metal area, dispersion, crystallite size			
Active site surface concentration			
Reduction, Oxidation Temperatures			
Acid site strength distribution: Lewis/Brønsted acid site distribution			
Breakthrough Curve Measurement			
Activation Energy			
Optional Capabilities			
CryoCooler	-100°C to 1200°C		
Detection by Mass Spec			
Continuous or Pulsed Vapor Dosing: water, alcohol, amines, aromatic organics, and more			
Enhanced Chemical Resistance			
B.E.T. Surface Area			

OPTIONSAND ACCESSORIES

MASS SPECTROMETER

Mass spec provides a direct probe for the identity and quantity of specific reaction products.

This is especially valuable when investigating an unknown reaction, or a reaction that creates multiple products. The single quadrupole mass spec with heated transfer line provides detection of mass fragments up to 200 amu and data collection that is integrated with the operation of the AutoChem III.

The AutoChem III also includes a general mass spec communication port for coordination with a lab's existing mass spec.

CRYOCOOLER

Begin experiments at temperatures as low as -100 °C with controlled liquid-nitrogen-based cooling.

VAPOR GENERATOR

Prepare samples for analysis or perform measurements in the presence of pulsed or continuous vapor streams such as water, alcohol, pyridine, aromatic organics, and more.

ENHANCED CORROSION RESISTANCE (ECR)

For reaction chemistries that requires particularly aggressive gas compositions, a special version of the AutoChem III is available with enhanced corrosion resistance. Wetted materials are constructed from highly resistant Hastelloy, highly stable perfluoroelastomers, and inert-coated stainless steel to provide the greatest stability under the harshest working conditions.



RELEVANTAPPLICATIONS



NET ZERO TECHNOLOGIES

The development of efficient and effective catalysts is necessary to the continued development of CO_2 mitigation and the hydrogen economy that will enable a sustainable energy future. The AutoChem III is a useful tool to optimize adsorption and dissociation of H_2/O_2 on electrolysis electrodes, show whether desorption occurs near reaction conditions, quantify acid or base sites to optimize reactivity and selectivity, and more.



FUEL CELLS

Platinum-based catalysts including Pt/C, PtRu/C, and PtRulr/C are often characterized by temperature-programmed reduction to determine the number of oxide phases and pulse chemisorption to calculate: metal surface area, metal dispersion, and average crystallite size.



PARTIAL OXIDATION

Manganese, cobalt, bismuth, iron, copper, and silver catalysts used for the gas-phase oxidation of ammonia, methane, ethylene, and propylene are characterized using: Temperature-programmed oxidation and desorption, heat of desorption & dissociation of oxygen.



CATALYTIC CRACKING

Acid catalysts such as zeolites are used to convert large hydrocarbons to gasoline and diesel fuel.

The characterization of these materials includes: Ammonia chemisorption and temperature-programmed desorption.



CATALYTIC REFORMING

Catalysts containing platinum, rhenium, tin, etc. on silica, alumina, or silica alumina are used for the production of hydrogen, aromatics, and olefins.



ISOMERIZATION

Catalysts such as small-pore zeolites (mordenite and ZSM-5) containing noble metals (typically platinum) are used to convert linear paraffins to branched paraffins.



HYDROCRACKING: HYDRODESULFURIZATION, AND HYDRODENITROGENATION

Hydrocracking catalysts typically composed of metal sulfides (nickel, tungsten, cobalt, and molybdenum) are used for processing feeds containing polycyclic aromatics that are not suitable for typical catalytic cracking processes.



WATER GAS SHIFT REACTION

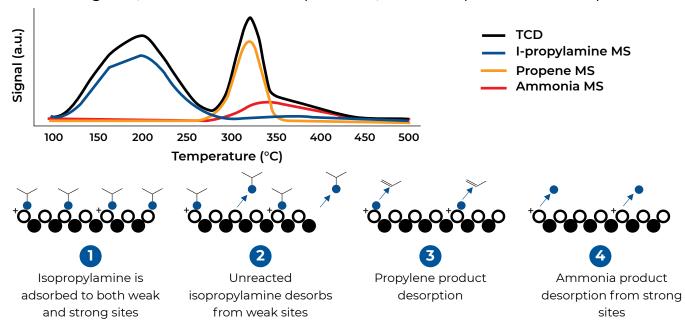
The water gas shift reaction is an important element in the hydrogen lifecycle and the push toward net zero technologies. The combination of catalysts, often copper-zinc-alumina and iron-chromia, are characterized by TPR and pulse chemisorption maximize activity.

16 APPLICATIONS APPLICATIONS

AUTOCHEMMETHODS

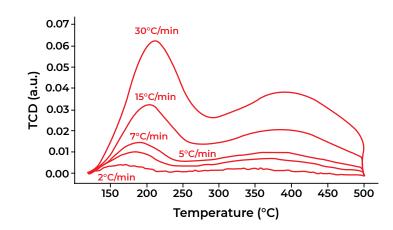
TEMPERATURE PROGRAMMED REACTIONS

The suite of temperature programmed reactions are collectively used to measure reactivity as a function of temperature through change in gas stream composition. As temperature is increased, gas stream composition is changed through: consumption of reactive gases, creation of reaction products, and desorption of bound species.



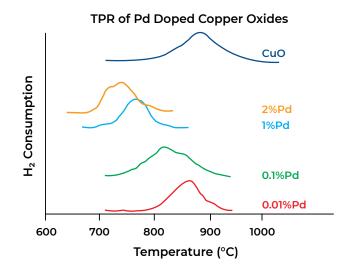
TEMPERATURE-PROGRAMMED DESORPTION (TPD)

Species previously adsorbed can be desorbed by increasing sample temperature under flowing inert gas. Ammonia TPD is one of the most common applications. A sample is first saturated with ammonia during the preparation step, then heated to desorb bound ammonia, revealing relative acid site strength for materials such as zeolites. Similarly, desorption of carbon dioxide indicates the strength of basic sites.



Bulk decomposition into the gas phase can also be used to characterize carbonates for CO₂ removal or hydrides for hydrogen storage.

TEMPERATURE PROGRAMMED REDUCTION (TPR)



The TPR measurement is a specific case of temperature programmed reaction in which a gas blend of hydrogen and an inert carrier – usually Argon – are passed over an oxide sample. Hydrogen is removed from the gas stream and water vapor is generated. The water vapor is trapped by the AutoTrap and the depletion of hydrogen from the carrier stream is measured.

In particular, this measurement provides the required conditions (temperature, time, and activation energy) required to prepare a heterogeneous catalyst from its native oxide state into the active zero-valence metal.

TEMPERATURE PROGRAMMED OXIDATION (TPO)

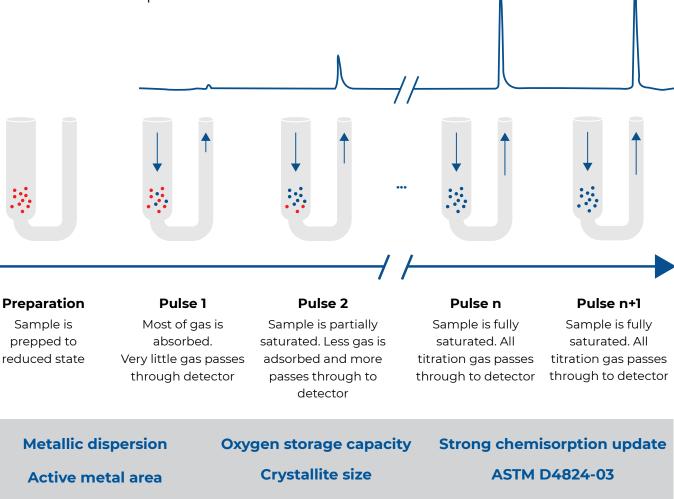
In a TPO experiment, oxygen in the sample gas stream is consumed, generally reacting with different forms of carbon to produce CO or CO₂. TPO experiments are important to characterize the **reactivity of metal oxide catalysts** and optimize process conditions.

The temperature of oxidation is also associated with the reactivity of carbon-carbon bonds, and is therefore an effective means for **differentiating forms of carbon**. Using TPO, amorphous, nanotube, filament and graphitic carbon can be distinguished, especially those that form on catalysts.

AUTOCHEMMETHODS

PULSE CHEMISORPTION

A sample is prepared in situ to a known initial state (e.g. pure oxide or valence metal) through temperature and gas exposure. Reactive gas pulses of known volume are delivered through to the sample and the system measures the volume of gas consumed in each pulse.

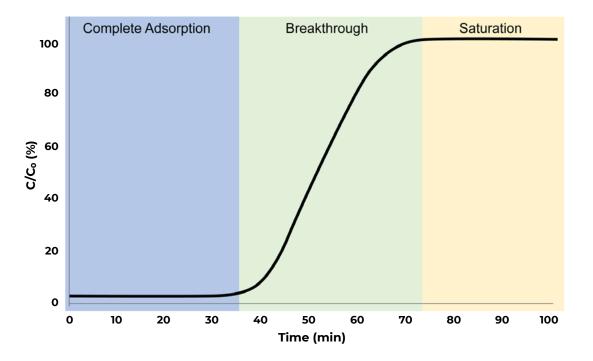


B.E.T. SURFACE AREA: PHYSISORPTION

The AutoChem III can measure B.E.T surface area by the flowing or dynamic method in which N_2 depleted from, or added to, a flowing gas stream is measured at liquid nitrogen, or ambient, temperatures, respectively. Basic physical surface area measurements are important for catalyst development as they reflect the basic physical form of the catalyst and/or support and the available contact area for reactivity. This is also an important basic measurement for porous and granular materials of all types.

BREAKTHROUGH CURVE ANALYSIS

Breakthrough analysis is a powerful technique for determining the adsorption capacity of a material under dynamic flow conditions. Breakthrough analysis allows users to precisely control temperature, pressure, and gas flowrates during an experiment. This allows users to analyze adsorbates under process relevant conditions, giving them the tools necessary to optimize their systems and adsorbent materials for their application. Additionally, breakthrough allows users to easily collect multicomponent equilibrium adsorption data allowing them to determine the selectivity and adsorption kinetics of their materials.





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Micromeritics' team provides industry-leading, high-quality application support and training to assist scientists, engineers, and analysts in the field of material characterization. Our application support team is composed of scientists and engineers to help users obtain the highest quality data and information about their material from Micromeritics Instruments. The Micromeritics team is dedicated to helping users successfully use their Micromeritics Instruments for the life of their instrument.



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Collaborations with industrial and academic partners to continually improve the quality of measurements and interpretation of material characterization data



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- Over 25 analytical techniques

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